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## DICTYOPTERENE A, AN ODORIFEROUS CONSTITUENT

## FROM ALGAE OF THE GENUS DICTYOPTERIS\*

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We wish to report the isolation and characterization of a novel hydrocarbon, dictyopterene A (1), from the essential oil of algae of the genus <u>Dictyopteris</u> (1, 2, 3).

Two species of <u>Dictyopteris</u>, <u>viz</u>. <u>D. plagiogramma</u> (Montagne) Vickers and <u>D. australis</u> Sonder, grow abundantly at times on the reefs of Waikiki, Honolulu. The fresh wet algae (unseparated) were placed in a large vacuum desiccator and the essential oil was trapped on the finger of a dry ice condenser. The volatile organic material was extracted with carbon disulfide, the carbon disulfide removed by fractional distillation, and the essential oil obtained in 0.005% yield by a short-path distillation, b.p. 90-150°. Separation of the components of the volatile oil was accomplished by preparative gas chromatography on a 10' x 3/8" column of 10% SE-30 on Aeropak 30. The first major compound to emerge from the gas chromatograph (Varian Aerograph Autoprep Model 705 with hydrogen flame ionization detector) was dictyopterene A:  $[\alpha]_{\rm D}^{21} + 77^{\circ} \pm 5^{\circ}$  (C 0.5, EtOH); mol wt 150 which corresponds to the empirical formula  $C_{11}H_{18}$ by mass spectrometry;  $\lambda_{\rm max}^{\rm EtOH}$  206 mu ( $\epsilon$  16,000).

A complete analysis of its nmr spectrum, achieved only with the aid of several double resonance experiments, led us to the conclusion that dictyopterene A is <u>trans</u>-l-(<u>trans</u>-l-hexenyl)-2-vinylcyclopropane (1)

<sup>\*</sup>Presented in part at the 155th National Meeting of the American Chemical Society, San Francisco, California, March 31 - April 5, 1968.



The nmr spectral data is summarized in Table I.

## TABLE I

Nmr Spectral Data of Dictyopterene A

Description of signal	No. of protons	Chemical shift <sup>a</sup>	Coupling constant <sup>b</sup>	Assignment
Broad 1:3:3:1 quartet <sup>c,d</sup>	2	1.95	7	-CH <sub>2</sub> - <u>CH</u> <sub>2</sub> -CH=
Doublet of triplets <sup>e</sup>	1	5.45	15.5 <sup>f</sup> , 7	$-CH_2$ $c = c < H$
Doublet of doublets of triplets <sup>g</sup>	1	4.99	15.5, 6, 1.2	$\frac{-CH_2}{H}$ $c = c < \frac{H}{CH_2}$
Doublet of doublets	1	4.98	17 <sup>f</sup> , 2.5 <sup>h</sup>	$rac{1}{rac} = c < \frac{H}{H}$
Doublet of doublets	1	4.81	10 <sup>1</sup> , 2.5	$rac{H}{R}$ c = c $\left< \frac{H}{H} \right>$
Doublet of doublets	1	5.37	17, 10, 6	$-c_{H}^{-c_{H}} c = c_{H}^{H}$
Triplet <sup>k</sup>	3	0.77	7	CH3-CH2-
Multiplet	6	<u>ca</u> . 1.25		$-\underline{CH}_2-\underline{CH}_2-\text{ and } \underset{H}{\overset{H}{\longrightarrow}}$
Triplet <sup>k</sup> , 1	2	0.87	7	

<sup>a</sup>Determined on a Varian HA-100 instrument in deuteriochloroform and reported as  $\delta$  units relative to TMS ( $\delta$  = 0).

<sup>b</sup>In cps.

<sup>c</sup>Collapses to a broad triplet (J = 7 cps) when irradiated at 5.45 ppm and to an unresolved doublet (J = 7 cps) when irradiated at 1.25 ppm.

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dThe signal is broad due to virtual coupling [see J. I. Musher and E. J. Corey, <u>Tetrahedron</u>, 18, 791 (1962)] to the  $\epsilon$ -methylene protons and long-range coupling (J = 1.2 cps) to the  $\alpha$ -methine proton of the hexenyl group.

 $^{e}$ Collapses to a doublet (J = 15.5 cps) when irradiated at 1.95 ppm.

f<sub>Jtrans</sub>.

8Collapses to a doublet of doublets (J = 15.5 and 6 cps) when irradiated at 1.95 ppm and to a doublet of triplets (J = 15.5 and 1.2 cps) when irradiated at 1.25 ppm.

h<sub>Jgem</sub>.

i<sub>Jcis</sub>.

 $\mathbf{j}$ Collapses to a doublet of doublets (J = 17 and 10 cps) when irradiated at 1.25 ppm.

 $k_{Collapses}$  to a singlet when irradiated at 1.25 ppm.

<sup>1</sup>The coincidence of chemical shifts for the two methylene protons establishes the <u>trans</u> configuration for the disubstituted cyclopropane ring.

Osmium tetroxide-periodate oxidation (4) of dictyopterene A yielded formaldehyde, nvaleraldehyde, and <u>trans</u>-1,2-cyclopropanedicarboxaldehyde that were converted to the corresponding 2,4-dinitrophenylhydrazones and separated by preparative thin layer chromatography on deactivated silica gel (1:1 isooctane-benzene). The DNP derivatives of formaldehyde and nvaleraldehyde were identical with authentic samples (mass spectra,  $R_f$ -values, and melting points). The <u>bis</u>-DNP of the cyclopropane dialdehyde was not obtained entirely pure, but had the expected molecular weight by mass spectrometry.

Treatment of dictyopterene A with osmium tetroxide followed by periodate-permanganate oxidation (5) afforded formic acid, n-valeric acid, and (+)-<u>trans</u>-1,2-cyclopropane dicarboxylic acid (6). Formic and valeric acids were identified by paper chromatography.

These data establish structure 1 unequivocally for dictyopterene A.

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